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A Compact CO₂ Amplifier Optically Pumped

by a Tunable 4.3 μm Fe:ZnSe Laser

A thesis submitted in partial satisfaction of the requirements for the degree of Master of Science in Electrical & Computer Engineering

by

Dana Thomas Tovey

ABSTRACT OF THE THESIS

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Master of Science in Electrical & Computer Engineering University of California, Los Angeles, 2021 Professor Chandra J. Joshi, Chair

There is currently an ongoing worldwide effort to develop long-wave infrared (8-14 μ m) sources with GW-level peak powers and ~1 kHz repetition rate for HHG generation as well as TW-level peak powers for particle acceleration and for long-distance atmospheric propagation. CO₂ lasers can store a great deal of energy and have been demonstrated to amplify 9-10 μ m picosecond pulses to very high peak powers. However, the electric discharge typically used for pumping limits these systems to low repetition rates and small aperture beams because of the complexity of the large-scale high-voltage devices used for pumping and the difficulty in maintaining a stable discharge across large apertures at the high pressures (>10 atm) required for short pulse amplification. In principle, optically pumped CO₂ lasers offer a compact alternative without these limitations, but the historic lack of high-energy pump sources at the appropriate wavelength has inhibited the study of picosecond pulse amplification in such a gain medium.

In this thesis, we first present a detailed analysis of 10 μ m lasing and gain dynamics in a CO₂ medium optically pumped at ~4.3 μ m by a continuously tunable Fe:ZnSe laser system. We then show that lasing can be achieved in a 6 cm long CO₂-He cell at total pressures up to 15 atm by tuning the pump wavelength far from the peak of CO₂ absorption. The optimization of experimentally measured CO₂ vibrational temperatures allows optical-to-optical conversion efficiencies of up to 30% to be reached at atmospheric pressures, falling to ~5% at pressures above 10 atm. At these high pressures, the gain lifetime is measured to be ~1 μ s, indicating the possibility of building both multi-pass and regenerative amplifiers using an optically pumped CO₂ medium. Numerical simulations based on density matrix formalism confirm that the amplification of a 3 ps pulse and a sub-picosecond pulse to GW-level powers is feasible in such a compact high-pressure optically pumped CO₂ amplifier.

The thesis of Dana Thomas Tovey is approved.

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University of California, Los Angeles

2021

Dedication

To my family and friends, thank you for your unwavering support and interest.

Table of Contents

1	Int	roduction	1
2	Th	eory	5
3	Fe	:ZnSe pump laser	12
4	Op	otically pumped CO ₂ active medium	14
	4.1	Low-pressure lasing	16
	4.2	Gain dynamics	18
	4.3	High-pressure lasing	23
	4.4	Gain lifetime	27
	4.5	Lasing in indirectly pumped CO ₂	29
	4.6	Possibility of repetition rate scaling	30
5	Sin	nulation results	32
	5.1	Direct amplification of a 3 picosecond pulse	35
	5.2	Chirped-pulse amplification of a sub-picosecond pulse	36
6	Co	nclusion	38

List of Figures

1	The CO ₂ gain spectrum at total pressures of 1 atm (blue) and 20 atm (gray). The black			
	curve indicates the bandwidth of a transform-limited 1 ps pulse centered at 10.28 μ m 1	ļ		
2	Optical pumping scheme for (a) direct excitation of CO ₂ and (b) indirect excitation of			
	CO ₂ via CO)		
3	A simplified energy band diagram of the CO2 molecule, with the three vibrational			
	modes represented by columns as labeled)		
4	The gain envelope of the regular 10 μ m P branch of CO ₂ transitions at a translational			
	temperature of 300 K. Red diamonds indicate the rovibrational lines on which small			
	signal gain was measured in experiment	3		
5	(a) Normalized population of CO ₂ molecules in each energy level of the asymmetric			
	mode when calculated with a Treanor distribution function (in red) and a Boltzmann			
	distribution function (in blue) for T_3 temperatures of 1800 K (top) and 4200 K			
	(bottom). (b) The normalized gain spectrum for a gas mixture of 1 atm CO_2 and 19 atm			
	He, optically pumped to a T ₃ value of 4200 K 10)		
6	(a) A simplified schematic of the Fe:ZnSe MOPA system. (b) The Fe:ZnSe master			
	oscillator)		
7	(a) Output energy of the Fe:ZnSe MOPA as a function of wavelength for the (i) master			
	oscillator, (ii) MOPA system at full pump energy without absorbing element between			
	MO and PA, (iii) MOPA system at full pump energy with absorber, and (iv) MOPA			
	system at reduced pump energy without absorber. (b) Typical temporal pulse profiles			
	for (i) the 2.94 μm Er:YAG pump laser, (ii) the master oscillator, and (iii) the MOPA			
	output	;		

8 Experimental setup of the optically pumped CO₂ laser. DM – dichroic mirror, the left 9 (a) The amount of 10 μ m energy generated in the optically pumped CO₂ laser as a function of absorbed pump energy. A least squares linear fit to the aggregate data is shown in black. (b) Optical-to-optical conversion efficiency in the optically pumped CO_2 laser for a pump wavelength of 4.23 μ m. The black curve indicates a least squares 10 (a) The asymmetric stretching mode vibrational temperature T_3 and (b) the translational temperature T within a cell of 50 torr pure CO₂ optically pumped at 4.3 µm as a function of time. The blue curves show the results of raw data; the red curves show this data with a 50 point moving average filter applied to reduce noise. The pump pulse temporal 11 (a) The asymmetric stretching vibrational mode temperature T_3 and (b) the corresponding peak gain coefficient in an optically pumped CO₂ active medium vs. absorbed 4.3 µm pump energy per volume. Each color corresponds to a different 12 The peak asymmetric stretching vibrational mode temperature, T₃, in an optically pumped CO₂ active medium as a function of total pressure for pure CO₂ (see blue 13 (a) Normalized absorption in 0.75 atm CO₂ and 9.25 atm He. Red markers indicate pump wavelengths used in experiment. (b) Maximum lasing pressure vs estimated experimental absorption coefficient. Numbers correspond to pump wavelengths

	indicated in (a). The green star indicates maximum lasing pressure after increasing the	
	pump intensity by a factor of ~ 2 to ~ 5.3 MW/cm ²	. 24
14	Beam profiles of (a) the 4.4 μm pump laser and (b) the 10.6 μm optically pumped CO_2	
	laser output at a total pressure of 15 atm.	. 25
15	(a) Optical-optical conversion efficiency of the optically pumped CO ₂ laser as a	
	function of pressure using three different output couplers. (b) Conversion efficiency as	
	a function of cavity Q factor at three different pressures. (c) 10 μm lasing energy	
	generated vs. 4.40 μ m pump energy absorbed at 7 atm total pressure.	. 26
16	The 4.40 μ m pump pulse (blue) and the amplification factor of the 10.6 μ m probe pule	
	(red) as a function of time for a gas mix of 0.75 atm CO_2 and 10.25 atm Helium	. 27
17	(a) Total amplification of the 10.6 μ m probe pulse in the 6 cm optically pumped CO ₂	
	cell and (b) gain lifetime as a function of total pressure	. 28
18	cell and (b) gain lifetime as a function of total pressure.(a) Normalized absorption in 1 atm CO. Red markers indicate pump wavelengths used	. 28
18		. 28
18	(a) Normalized absorption in 1 atm CO. Red markers indicate pump wavelengths used	
	(a) Normalized absorption in 1 atm CO. Red markers indicate pump wavelengths used in experiment. (b) Maximum lasing pressure vs estimated experimental absorption	
	 (a) Normalized absorption in 1 atm CO. Red markers indicate pump wavelengths used in experiment. (b) Maximum lasing pressure vs estimated experimental absorption coefficient. Numbers correspond to pump wavelengths indicated in (a) 	
	 (a) Normalized absorption in 1 atm CO. Red markers indicate pump wavelengths used in experiment. (b) Maximum lasing pressure vs estimated experimental absorption coefficient. Numbers correspond to pump wavelengths indicated in (a) (a) Deflected probe signal (in blue) and Fe:ZnSe pump pulse (in red). (b) Deflected 	. 30
19	 (a) Normalized absorption in 1 atm CO. Red markers indicate pump wavelengths used in experiment. (b) Maximum lasing pressure vs estimated experimental absorption coefficient. Numbers correspond to pump wavelengths indicated in (a) (a) Deflected probe signal (in blue) and Fe:ZnSe pump pulse (in red). (b) Deflected probe signal and recovery. The dashed white line indicates an exponential fit with a 	. 30
19	 (a) Normalized absorption in 1 atm CO. Red markers indicate pump wavelengths used in experiment. (b) Maximum lasing pressure vs estimated experimental absorption coefficient. Numbers correspond to pump wavelengths indicated in (a) (a) Deflected probe signal (in blue) and Fe:ZnSe pump pulse (in red). (b) Deflected probe signal and recovery. The dashed white line indicates an exponential fit with a time constant of 0.33 ms. 	. 30
19	 (a) Normalized absorption in 1 atm CO. Red markers indicate pump wavelengths used in experiment. (b) Maximum lasing pressure vs estimated experimental absorption coefficient. Numbers correspond to pump wavelengths indicated in (a) (a) Deflected probe signal (in blue) and Fe:ZnSe pump pulse (in red). (b) Deflected probe signal and recovery. The dashed white line indicates an exponential fit with a time constant of 0.33 ms. (a) The gain spectrum of an active medium comprised of 1 atm CO₂ (63% ¹²C¹⁶O₂ 	. 30
19	 (a) Normalized absorption in 1 atm CO. Red markers indicate pump wavelengths used in experiment. (b) Maximum lasing pressure vs estimated experimental absorption coefficient. Numbers correspond to pump wavelengths indicated in (a) (a) Deflected probe signal (in blue) and Fe:ZnSe pump pulse (in red). (b) Deflected probe signal and recovery. The dashed white line indicates an exponential fit with a time constant of 0.33 ms. (a) The gain spectrum of an active medium comprised of 1 atm CO₂ (63% ¹²C¹⁶O₂ isotopologue, 37% ¹³C¹⁶O₂) and 19 atm He, pumped to a T₃ value of 3300 K. The 	. 30

	$(21\% {}^{12}C^{16}O_2, 43\% {}^{12}C^{18}O_2, and 36\% {}^{12}C^{16}O^{18}O$ isotopologue) and 19 atm He. A total	
	bandwidth of \sim 3 THz centered around 9.3 μ m wavelength is achieved	33
21	Simulation results for the amplification of a 3 ps pulse centered at 10.29 μ m in a gas	
	mix of 1 atm CO ₂ and 19 atm He optically excited to a T_3 value of 4200 K. (a) The	
	temporal profile of the pulse before (top) and after (bottom) amplification. (b) The	
	pulse energy as a function of amplifier length, discounting losses due to propagation or	
	output coupling	36
22	Simulation results for the amplification of a sub-ps pulse in an optically pumped CO ₂	
	amplifier comprised of 1 atm CO2 (63% 12CO2, 37% 13CO2) and 19 atm He. The	

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1 Introduction

Picosecond and subpicosecond pulses with TW-level power in the long-wave infrared region (8-14 μ m) are desirable for the study of high-field physics and nonlinear optics in this region of high atmospheric transmission [1-4]. At present, high-pressure CO₂ laser systems are the only viable candidate for generating such pulses, because the CO₂ molecule is capable of storing tens of Joules of energy in a large volume for 10 μ m amplification and does not face the damage threshold limitations that inhibit an optical parametric amplifier from reaching high peak powers at these wavelengths. In addition, while the individual rovibrational transitions of the CO₂ gain spectrum have a relatively narrow bandwidth, these transitions can be broadened via collisions at high pressures to overlap with one another and provide the THz bandwidth necessary for picosecond pulse amplification (see Figure 1). CO₂ lasers have been used to amplify 10 μ m pulses as short as a few ps [5-7]. Recently, powers as high as 15 TW have been achieved using CO₂ lasers in a master-oscillator power-amplifier (MOPA) system at the UCLA Neptune Laboratory [6], and 5

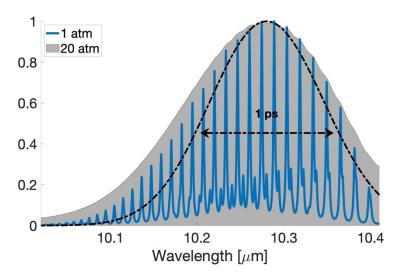


Figure 1. The CO₂ gain spectrum at total pressures of 1 atm (blue) and 20 atm (gray). The black curve indicates the bandwidth of a transform-limited 1 ps pulse centered at 10.28 μ m.

TW, 2 ps pulses have been generated using CO₂ amplifiers at the Accelerator Test Facility (ATF) at Brookhaven National Laboratory [7]. CO₂ laser systems are typically pumped with an electric discharge, however, and the voltage required for breaking the gap between electrodes scales linearly with pressure. As a result, at the high pressures (>10 atm) needed for a smooth broadband gain spectrum, it is extremely difficult to maintain a stable electric discharge in large volumes, and repetition rates are limited.

In principle, the CO₂ active medium can be excited optically to circumvent this discharge pressure limitation. As shown in Figure 2(a), a pump source at ~4.3 μ m can be used to directly populate the upper laser level 001 of the 10 μ m laser channel and excite the entire asymmetric stretching mode of the CO₂ molecule, while the lower laser level 100 remains only weakly thermally populated. Optical pumping of a CO₂ laser was historically first demonstrated using an incoherent CO-air flame to excite CO₂ at ~4.3 μ m, producing ~1 mW of cw radiation at 10.6 μ m [8]. Later, several groups explored the potential of optically pumping a high-pressure CO₂ laser, which would exhibit continuous spectral tuning across the broad gain bandwidth. One such group demonstrated lasing in optically pumped CO₂ at a total pressure of 33 atm using a pulsed ~4.23

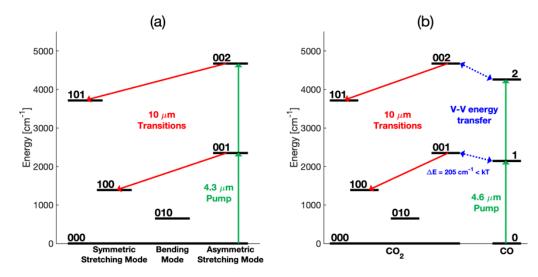


Figure 2. Optical pumping scheme for (a) direct excitation of CO_2 and (b) indirect excitation of CO_2 via CO.

 μ m HBr laser for pumping, but optical-to-optical conversion efficiency was not optimized or reported [9]. In addition, the gain length was restricted to only ~1 mm due to the extremely high absorption coefficient and short gain lifetime at such high pressures. This reduced the cavity roundtrip time but also decreased the single-pass gain in such a short CO₂ cell. The 101 level of the CO₂ molecule has also been pumped using a ~2.7 μ m HF chemical laser [10,11], but this method was not pursued beyond the initial experimental demonstration, as the fast collisional decay of the 101 level simultaneously populates both the upper (001) and lower (100) laser levels for the 10 μ m band. In general, the lack of available energetic pump sources operating at wavelengths that are efficiently absorbed by CO₂ has prevented the demonstration of high gain in a high-pressure optically pumped CO₂ amplifier.

An alternative optical pumping scheme has also been explored, in which a partner molecule is resonantly excited by optical pumping and the absorbed energy is then transferred to CO_2 via collisions. This method enables the use of pump wavelengths that do not overlap directly with CO_2 absorption lines and can also result in significantly longer gain lifetimes due to the slow exchange of vibrational energy between different molecules such as N₂O and CO. Figure 2(b) depicts such a scheme using CO as the collisional partner molecule to be optically excited. The small quantum defect between the first vibrational level of CO and the first level of the CO₂ asymmetric stretching mode allows energy to be efficiently transferred from CO to CO₂. Lasing was demonstrated in a CO-CO₂-He mix at a total pressure of 16 atm by pumping CO at 4.8 µm using the second harmonic of a 9.6 µm TEA CO₂ laser [12]. Less than 1 mJ of 10 µm light was generated, however, as this method has faced a similar lack of energetic pump sources.

Recently, significant progress has been made on solid-state lasers using iron-doped zinc chalcogenides to provide tunable sources around 3-5 μ m that have the potential to be scaled up to

Joule-class energies [13]. Specifically, the development of an Fe:ZnSe laser system capable of producing ≤ 60 mJ, 200 ns pulses at a wavelength tunable from 3.8-5.0 µm has provided an energetic pump source for optically exciting CO₂ either directly at ~4.3 µm or indirectly by pumping CO at ~4.6 µm [14]. This has opened a new opportunity for studying the feasibility of generating or amplifying picosecond or sub-picosecond 10 µm pulses in an optically pumped CO₂ active medium, which has never before been thoroughly analyzed.

In this thesis, we present a detailed study of lasing and gain dynamics in a CO₂ active medium optically pumped by such a tunable Fe:ZnSe laser. The thesis will be structured as follows. Chapter 2 will introduce a generally accepted theoretical temperature model that describes an excited CO₂ medium, providing necessary background information for understanding the gain dynamics of a CO₂ laser. Chapter 3 will detail our experimental results, including the demonstration of high gain and high conversion efficiencies in optically pumped CO₂ at atmospheric pressures [15], the optimization of pump wavelength to achieve lasing at total pressures up to 15 atm, and measurements of gain lifetime as a function of pressure [16]. The alternative pumping scheme in which CO₂ is indirectly excited via pumping CO will also be discussed here [16]. We also evaluate experimentally a maximum repetition rate at which a CO₂ medium can be optically pumped by analyzing the dissipation of pressure or heat waves caused by the pump pulse [16]. Chapter 4 will detail theoretical simulations that model the amplification of short pulses in a high-pressure optically pumped CO₂ amplifier [17]. Finally, Chapter 5 will present our conclusions and look forward to future work.

2 Theory

Ultimately, the development of an optically pumped CO₂ laser system capable of generating high power 10 µm pulses requires a detailed understanding of gain dynamics in such an optically excited molecular active medium. The properties of a discharge pumped CO₂ gain medium have been well characterized using a generally accepted temperature model in which each vibrational mode is assigned a vibrational temperature describing the Boltzmann distribution of population in that mode [18-22]. This model can similarly be applied to an optically pumped CO₂ system, as the relationship between the relaxation times of the main kinetic processes occurring in molecular gases remains the same:

$$\tau_{VR} \ll \tau_{VV} \ll \tau_{VV'} \tag{1}$$

where τ_{VR} , τ_{VV} , and $\tau_{VV'}$ are time constants describing the rates at which equilibrium is achieved among rotational energy levels, among vibrational energy levels within a single vibrational mode, and among different vibrational modes of the CO₂ molecule, respectively. Equilibrium is reached among individual rotational levels first – such that the population among these levels can be described according to a Boltzmann distribution with translational or gas temperature, T – on a time scale of $\tau_{VR} \approx 0.2$ ns at 1 atm of pressure [18]. The distribution of population within the different vibrational energy levels in a single vibrational mode reaches a Boltzmann distribution on a time scale of $\tau_{VV} \approx 5$ ns at 1 atm [18]. Equilibrium among different vibrational modes or different gases (e.g. N₂) within a mixture occurs on a much longer time scale of $\tau_{VV'} \leq 1 \ \mu s$ at 1 atm [18]. This hierarchy of relaxation times in the CO₂ molecule is the main reason for the applicability of this temperature model when considering a typical pump laser pulse on the order of 100 ns. The population of each vibrational energy level can thus be accurately described using a Boltzmann distribution with a different temperature, T_i , assigned to each vibrational mode *i*:

$$N_{\nu} = N_0 \exp\left(-\frac{hc\nu G_i}{k_B T_i}\right) \tag{2}$$

where N_{ν} is the population of molecules in level ν of the mode, N_0 is the population of molecules in the ground state, G_i is the energy gap between levels in vibrational mode *i*, and *h*, *c*, and k_B are the usual physical constants.

The CO₂ molecule has three vibrational modes: symmetric stretching (T_1) , bending (T_2) , and asymmetric stretching (T_3) . Figure 3 shows the vibrational energy levels of interest in a CO₂ laser, with each column corresponding to a distinct vibrational mode. It can be seen that the upper laser levels for 10 µm lasing transitions in CO₂ are energy states corresponding to excitation of the

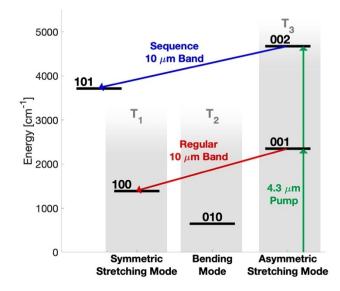


Figure 3. A simplified energy band diagram of the CO₂ molecule, with the three vibrational modes represented by columns as labeled.

asymmetric stretching mode. As a result, optimizing gain in a CO₂ active medium corresponds to maximizing T_3 without increasing T_1 and T_2 . In the case of optical pumping with a ~4.3 µm pump source, it is reasonable to assume that all of the absorbed pump energy is stored in this asymmetric stretching mode, while the vibrational temperatures of other modes, T_1 and T_2 , tend to remain equal to the translational temperature of the gas, T. As a result, optimization of gain is easier to achieve in an optically pumped CO₂ active medium than in discharge-excited CO₂ systems [19]. This ability to selectively excite the asymmetric stretching mode via the absorption of 4.3 µm pump photons is an additional advantage of pumping CO₂ optically. It has been demonstrated that in an optically pumped CO₂ medium, T_3 values as high as 4200 K have been reached [23], while traditional discharge pumped CO₂ lasers fail to reach T_3 values above 1800 K [18].

For optical excitation, the population of molecules in each individual energy level can be determined with knowledge of the translational temperature, T, and this vibrational temperature, T_3 . Full characterization of an optically pumped CO₂ gain medium can thus be achieved by making time-resolved measurements of both T and T_3 . The translational temperature can be calculated from measurements of small signal gain on multiple different regular band transitions. Figure 4 shows the envelope of the 10P branch of the normalized CO₂ gain spectrum as a function of rotational quantum number j at an example temperature of T = 300 K. The shape of this envelope depends on the translational temperature since the population of molecules in a particular rotational-vibrational state vj is given by [18]:

$$N_{vj} = N_v \left(\frac{2hcB_v}{k_BT}\right) (2j+1) \exp\left(-\frac{hc}{kT} \left[B_v j(j+1) - D_v j^2 (j+1)^2\right]\right)$$
(3)

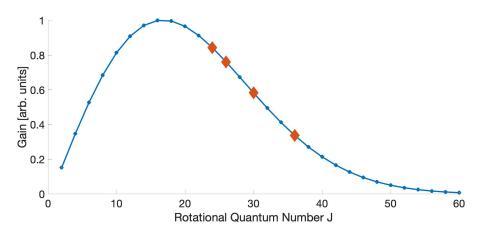


Figure 4. The gain envelope of the regular 10 μ m P branch of CO₂ transitions at a translational temperature of 300 K. Red diamonds indicate the rovibrational lines on which small signal gain was measured in experiment.

where *v* refers to the specific vibrational state (for example, 001), N_v is the population of molecules in the vibrational state as a whole (given in Equation 2), B_v and D_v are rotational constants for the particular vibrational state, and *j* is the rotational quantum number [23]. The translational temperature *T* can thus be found by fitting a curve to experimentally measured small-signal gain values on various 10P CO₂ lines. Analysis of this method of calculating *T* revealed that the slope of the curve on the high-*j* side of the peak of the gain distribution is very sensitive to changes in temperature. Chapter 4.2 will discuss our experimental measurements of *T* as a function of time using this method; we chose to measure gain on the 10P(24), 10P(26), 10P(30), and 10P(36) rovibrational transitions (see Figure 4) to maximize our temperature measurement accuracy. The accidental overlap between regular band lines (001-100) and sequence band (002-101) or hot band (011-110) lines was also considered when choosing rovibrational lines, as these overlaps can distort the gain envelope and introduce additional error in the measurement of *T* [24,25]. The asymmetric stretching mode temperature, T_3 , can be calculated from measurements of small signal gain on both a regular band transition and a sequence band transition using the following relation [18,23]:

$$\frac{g_{\text{seq}}}{g_{\text{reg}}} = 2 \exp\left(-\frac{hcG_3}{k_B T_3}\right) \tag{4}$$

which gives:

$$T_3 = \frac{hcG_3}{k_B \ln\left(\frac{2g_{\text{reg}}}{g_{\text{seq}}}\right)} \tag{5}$$

where g_{seq} is the small signal gain on a sequence band transition (002-101), and g_{reg} is the small signal gain on a regular band transition (001-100) with an equivalent rotational quantum number *j*. This equation, however, assumes a perfect Boltzmann exponential distribution of population in the asymmetric stretching mode of CO₂. Previous experiments have indicated that for the very high values of T_3 that can be achieved with optical pumping, the distribution of population in the asymmetric mode can be more accurately described using a Treanor distribution function [23,26]:

$$N_{00n} = N_0 \exp\left(-\frac{hc}{k} \left(\frac{nG_1}{T_3} - \frac{n(n-1)w_e x_e}{T}\right)\right)$$
(6)

where N_{00n} , N_0 , and G_1 are defined above, and $w_e x_e$ is the anharmonicity constant introduced in this formula. This arises from the fact that the Boltzmann distribution is based on the assumption that all vibrational levels in a single mode are evenly spaced. In reality, the separation between levels in the CO_2 asymmetric stretching mode is not constant but rather decreases for higher levels, resulting in a slightly less efficient exchange of energy between these levels and therefore a distribution of energy that is not perfectly exponential.

Calculations have been performed to determine whether this change in distribution function plays a significant role for an optically pumped CO₂ medium. Figure 5(a) displays the normalized population in each 00n energy level of CO₂ when calculated using a Boltzmann distribution and a Treanor distribution for T_3 values of 1800 K and 4200 K. Recall that these two values correspond to the respective temperatures that are typically reached when pumped by electric discharge and

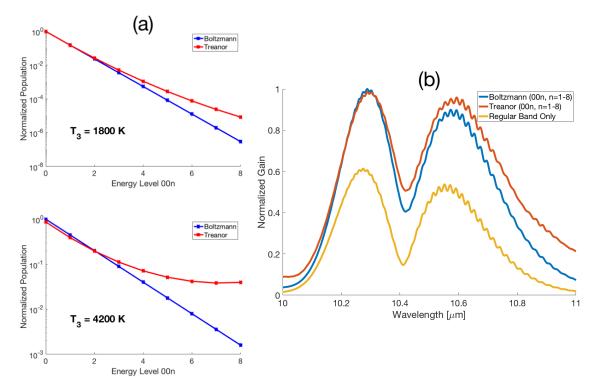


Figure 5. (a) Normalized population of CO_2 molecules in each energy level of the asymmetric mode when calculated with a Treanor distribution function (in red) and a Boltzmann distribution function (in blue) for T₃ temperatures of 1800 K (top) and 4200 K (bottom). (b) The normalized gain spectrum for a gas mixture of 1 atm CO_2 and 19 atm He, optically pumped to a T₃ value of 4200 K.

that can be reached when pumped optically. It is apparent that, in comparison with the Boltzmann distribution function, the Treanor distribution function places a significantly greater portion of the total population in the 00n energy levels for n > 3, increasing the gain present on upper sequence bands (00n-10(n-1)). It should be noted that while these bands are red-shifted due to anharmonicity, the transition frequencies of their R and P branches are relatively close to that of the regular band (<100 MHz difference) and are therefore still involved in the amplification process for the case of picosecond pulses. Figure 5(b), which displays the normalized gain spectrum around 10 µm for a 20 atm CO₂-He gas mix (containing 1 atm pure CO₂) that has been optically excited to a T_3 value of 4200 K, illustrates how sequence bands (00n-10(n-1)) contribute to the overall gain spectrum. The yellow curve represents gain on the regular band transitions, excluding extra gain from the sequence bands discussed above. The blue and red curves represent the gain spectrum with all sequence bands included when determining the asymmetric mode level populations using a Boltzmann distribution function and Treanor distribution function, respectively.

The yellow gain curve depicted in Figure 5(b) demonstrates that sequence band transitions have a significant contribution to the overall gain spectrum in a highly excited CO₂ gain medium. The red and blue curves show that the difference in peak gain at high pressures when calculated using a Treanor function in place of a Boltzmann function is negligible for realistic T_3 values at high pressures. It should be noted that the use of a Treanor function instead of a Boltzmann function to model population only substantially affects higher levels and sequence bands. As a result, using the Treanor function does slightly red-shift the wavelength at which gain is maximized, but the overlap between sequence band and regular band rovibrational lines at high pressures eliminates any significant disparity. Because differences in the gain spectrum at high pressures using different population distributions are negligibly small (see Figure 5(b)), the remainder of this thesis uses a Boltzmann function to describe population and to measure T_3 .

3 Fe:ZnSe pump laser

The experiments discussed in the following chapter of this thesis were performed using a Fe:ZnSe MOPA system for optically pumping a CO₂ gain medium [14]. A simplified schematic of this system is shown in Figure 6(a). A flashlamp-pumped, Q-switched 2.94 μ m Er:YAG laser producing ~200 ns pulses of ~250 mJ of energy is used to pump the system, with the 2.94 μ m pulse being split in multiple ways to pump both the oscillator and the three power amplifiers as shown. A diffraction grating within the Fe:ZnSe master oscillator laser cavity allows for continuous tuning over a wavelength range of 3.8-5.0 μ m. A picture of this master oscillator with the diffraction grating, Fe:ZnSe crystal, and Er:YAG pump laser labeled is shown in Figure 6(b). It should be noted that because the Fe:ZnSe gain spectrum is peaked at ~4.15 μ m, the gain medium will produce broadband amplified spontaneous emission centered at ~4.15 μ m when the master oscillator is tuned to operate at long wavelengths >4.6 μ m. As a result, while this did not affect

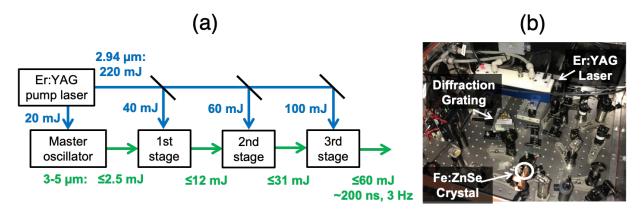


Figure 6. (a) A simplified schematic of the Fe:ZnSe MOPA system. (b) The Fe:ZnSe master oscillator.

our operation of the Fe:ZnSe laser for pumping CO₂ at ~4.3 μ m, studying the indirect pumping scheme in which CO is optically pumped at ~4.6 μ m required the introduction of an unpumped Fe:ZnSe crystal as an absorbing element between the master oscillator and power amplifier to eliminate this parasitic self-lasing [14].

Figure 7(a) shows measurements of pulse energy as a function of output wavelength under different pumping conditions. The black curve shows the tuning curve of the master oscillator, multiplied by 4 for visibility. The black, red, and blue markers show the tuning curve of the MOPA system at full pump energy without any absorbing element, at full pump energy with the absorbing element, and with reduced pump energy without any absorbing element, respectively. For the low pressure optically pumped CO₂ experiments discussed in Chapters 4.1-4.2, only the master oscillator with pulse energies of \leq 2.5 mJ was used for pumping CO₂, as the power amplifier system had not yet been fully developed [27]. For the high-pressure experiments discussed in Chapters

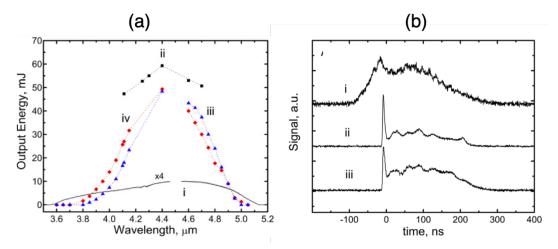


Figure 7. (a) Output energy of the Fe:ZnSe MOPA as a function of wavelength for the (i) master oscillator, (ii) MOPA system at full pump energy without absorbing element between MO and PA, (iii) MOPA system at full pump energy with absorber, and (iv) MOPA system at reduced pump energy without absorber. (b) Typical temporal pulse profiles for (i) the 2.94 µm Er:YAG pump laser, (ii) the master oscillator, and (iii) the MOPA output.

4.3-4.6, the full MOPA system was utilized, with the absorbing element put in place for the indirect pumping scheme results discussed in Chapter 4.5. Figure 7(b) shows typical temporal profiles for the 2.94 μ m Er:YAG pump laser, the master oscillator, and the entire MOPA system. Further details on this system can be found elsewhere [14].

4 Optically pumped CO₂ active medium

Experimental measurements of lasing and small-signal gain in an optically pumped CO₂ medium were all performed using a similar experimental setup, which is shown in Figure 8. The green arrows indicate the optical path of the Fe:ZnSe pump laser. A NaCl wedge angled at 45 degrees was used to sample ~1% of the pump pulse to diagnose the input pump energy prior to reaching the CO₂ cell as shown. For all experiments discussed in this chapter, the entire experimental area was purged with N₂ or Ar gas to eliminate any absorption of the pump pulse by CO₂ in the ambient air.

For lasing experiments, the laser cavity was formed by two curved dichroic mirrors surrounding a gas cell sealed with Brewster-angled NaCl windows. The dichroic mirrors are 99.5% reflective at ~10.6 μ m and 99% transmissive at ~4.3 μ m and have a radius of curvature of 70 cm. Low-pressure lasing measurements were performed using a 25 cm long cell; high-pressure lasing measurements where CO₂ absorption was stronger were performed using a 6 cm long cell. For all measurements, the cavity length was ~50 cm. The 10.6 μ m laser radiation decoupled through the downstream (leftmost in Figure 8) dichroic mirror was measured as shown. A 10 μ m narrow-bandpass filter was used to eliminate any residual 4.3 μ m pump radiation. As indicated in Figure 8, one of the dichroic mirrors could be replaced with a diffraction grating to vary the cavity Q

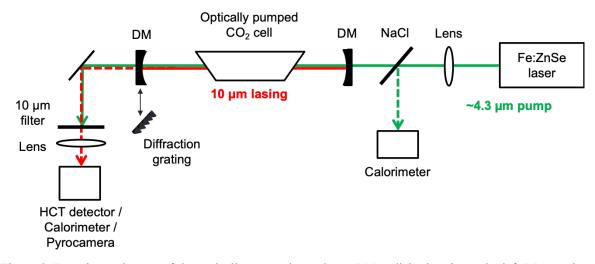


Figure 8. Experimental setup of the optically pumped CO_2 laser. DM – dichroic mirror, the left DM can be replaced by a diffraction grating.

factor. For this configuration, the zeroth order of the grating was used to decouple radiation out of the cavity. Two different diffraction gratings were used in this study, with measured reflectivities of \sim 71% and \sim 97.7% at 10.6 µm and groove densities of 135 gr/mm and 150 gr/mm, respectively.

To measure small-signal gain, the left dichroic mirror was removed, and a commercial, low-pressure, discharge-pumped, pulsed CO₂ laser was used to probe the active medium. Typical probe pulses contained a few mJs of energy and consisted of a short (~100 ns) spike followed by a long (>20 μ s) tail. Measurements were performed on the tail of the pulse such that the probing peak power was only a few Watts, far below gain saturation.

For the low-pressure gain measurements on both the regular and sequence bands described in Chapter 4.2, the 97.7% reflective diffraction grating was placed in a non-Littrow configuration within the probe laser cavity such that this probe laser pulse is doubly diffracted. The probe laser cavity length of ~ 1 m and this double passing through the diffraction grating increased dispersion to allow for precise tuning of the probe laser on individual rovibrational transitions of both the regular and sequence 10 µm branches of CO₂ [28]. Recall from Chapter 2 that measuring gain on several rovibrational transitions allows for time-dependent measurements of the vibrational mode temperatures in a CO₂ active medium. To maximize the Q-factor of the cavity, the zeroth-order reflection of the grating was used as an output coupler. Note that even with this scheme, only sequence lines with a high rotational quantum number (j > 21) were observed. This is a result of the frequency separation between adjacent CO₂ regular band and sequence band transitions increasing with rotational quantum number j such that the nearby regular band transition dominates lasing within the cavity for low j sequence band transitions. The probe pulse was sent through the optically pumped CO₂ cell in a double-pass scheme with the rightmost dichroic mirror used for reflection. A 50/50 beam-splitter directed the reflected pulse into a HgCdTe signal detector for measurement. Time-resolved gain could then be determined from comparison of the temporal profiles of the probe pulse and pump pulse, and, thus, time-resolved measurements of T_3 and T were obtained using the methods described previously in Chapter 2.

For the high-pressure measurements of gain lifetime discussed in Chapter 4.4, where precise tunability was no longer required, the intra-cavity grating was removed from the probe laser cavity and a 50% reflective mirror was used as an output coupler. To prevent cross-interaction between the pump and probe laser, the probe beam was aligned through the cell and reflected off the right dichroic mirror at a slight angle, measuring gain in a V-shaped double-pass scheme.

4.1 Low-pressure lasing

Our first experimental results were the demonstration of efficient lasing in an optically pumped CO_2 active medium at low pressures. Figure 9(a) shows the 10 µm energy generated from lasing in a cell containing 35 Torr of pure CO_2 optically pumped by ~2 mJ pulses generated by the Fe:ZnSe master oscillator. Here, the 97.7% reflective diffraction grating was used as an output

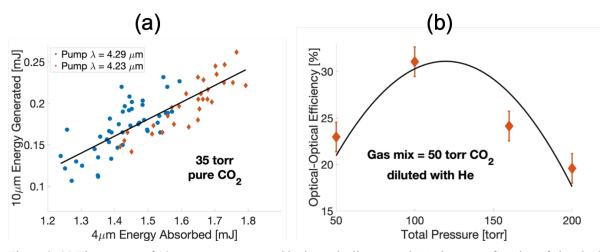


Figure 9. (a) The amount of 10 μ m energy generated in the optically pumped CO₂ laser as a function of absorbed pump energy. A least squares linear fit to the aggregate data is shown in black. (b) Optical-to-optical conversion efficiency in the optically pumped CO₂ laser for a pump wavelength of 4.23 μ m. The black curve indicates a least squares parabolic fit to the data.

coupler in place of the second dichroic mirror shown in Figure 8. The ~4.3 µm pump pulse energy was measured before the cell using the reflection off of the NaCl wedge and after the cell by measuring the pump energy reflected off of this diffraction grating. Blue dots correspond to lasing output when the pump laser was tuned to 4.29 µm, the peak of the 4P branch of the CO₂ absorption spectrum. Red dots correspond to a pump wavelength of 4.23 µm, the peak of the 4R branch, where absorption is naturally higher, allowing for more efficient pumping. A linear slope can be seen with no indication of pump energy saturation. Figure 9(b) shows the optical-to-optical conversion efficiency of 4.23 µm pump energy to 10.6 µm lasing energy as a function of total pressure in mixtures of 50 Torr of CO₂ and ballast He. A peak conversion efficiency of >30% was measured, nearing the theoretical quantum limit of ~40%.

Using this diffraction grating as an output coupler, lasing was not observed at pressures >1 atm. Using a second identical dichroic mirror to maximize the Q-factor of the cavity, however, allowed for the observation of lasing at total pressures up to 3 atm (50 Torr CO_2 diluted with He).

At such pressures, the collisional broadening of rovibrational lines resulted in stronger absorption of the pump energy, and it was thus necessary to tune the pump wavelength along the high-*j* transitions of the 4P absorption branch to ~4.35 μ m, where absorption is significantly lower. Stable lasing also required a precise tuning of the pump laser wavelength on the line center to avoid resonant nonlinear optical effects in CO₂, as self-focusing and -defocusing of the pump beam were observed when pumping CO₂ on the red side and the blue side of resonant rovibrational transitions, respectively [29]. It was discovered that both of these nonlinear effects negatively impacted the efficiency of the optically pumped CO₂ laser, and a cylindrical "pencil-like" pump beam was optimal.

Prior to the development of the MOPA system, the pump pulses provided by the master oscillator were limited to energies of ≤ 2.5 mJ, and it was estimated that this lack of pump energy limited the observation of high gain and lasing at pressures above 3 atm. To confirm this hypothesis, a more detailed study of the gain dynamics in optically pumped CO₂ was then carried out, which will be described in the following section.

4.2 Gain dynamics

As detailed in Chapter 2, the distribution of population and thus the gain dynamics of an excited CO₂ medium can be accurately described using a temperature model [20,21]. Knowledge of temperature kinetics is critical for laser optimization, and experimentally observed molecular kinetics in CO₂ lasers have been shown to deviate considerably from theoretical modeling, especially at high levels of vibrational excitation [20,23]. To study this, we performed small-signal gain measurements on different rovibrational lines of the CO₂ gain spectrum to experimentally

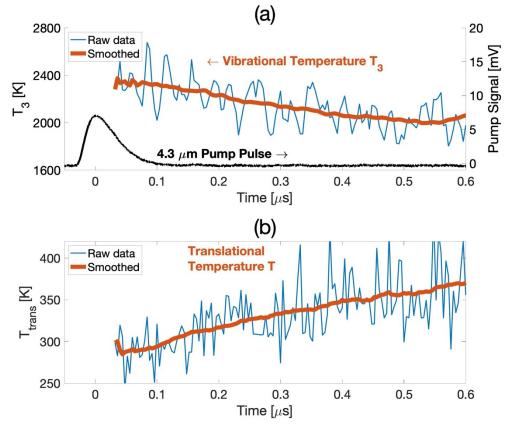


Figure 10. (a) The asymmetric stretching mode vibrational temperature T_3 and (b) the translational temperature T within a cell of 50 torr pure CO₂ optically pumped at 4.3 µm as a function of time. The blue curves show the results of raw data; the red curves show this data with a 50 point moving average filter applied to reduce noise. The pump pulse temporal profile is shown in black.

measure the vibrational temperature, T_3 , and the rotational temperature, T, as a function of time in a CO₂ cell optically pumped at ~4.3 µm using the methods described in Chapter 2 [15].

Figure 10 shows the results of these measurements for an active medium of 50 torr pure CO₂ pumped at 4.29 μ m. Asymmetric stretching mode vibrational temperature, T_3 , and translational temperature, T, are shown as a function of time. It can be seen that almost immediately after the absorption of the pump pulse (at time t = 0), a peak T_3 value of ~2400 K is reached, while translational temperature is measured to be near room temperature (~300 K) as expected. This T_3 value corresponds to a record-high peak gain coefficient of ~30%/cm. Note that

a TEA CO₂ laser pumped by electric discharge rarely achieves a T_3 value above ~1800 K or gain coefficients above 2-3%/cm [18]. Even optimized, non-self-sustained discharge-pumped CO₂ systems are typically only capable of reaching gain coefficients of 4-6%/cm. The asymmetric stretching mode then collisionally relaxes and energy is redistributed to other modes, resulting in a decrease in T_3 and an increase in T as shown in Figure 10. It should be noted the observation of gain on a sequence band transition (002-101), which was required for this measurement of T_3 , immediately after pumping gives further confirmation that the asymmetric stretching mode reaches equilibrium before the gas mixture as a whole.

These measurements were repeated for various CO₂-He mixtures to determine the effect of adjusting total pressure and CO₂ concentration on peak values of T_3 . Large T_3 values were only obtained in mixtures with small amounts of CO₂ gas, again indicating that greater pump energy is required to obtain high T_3 values in a larger partial pressure of CO₂. To study the scalability of this system to the high pressures required for sub-picosecond pulse amplification, theoretical calculations were performed to estimate the 4.3 µm pump energy required to achieve high gain in CO₂ at different pressures. Figure 11 shows theoretical calculations for the vibrational temperature, T_3 , and the corresponding peak gain coefficient as a function of absorbed energy per unit volume for various amounts of pure CO₂ gas. The pump energy is assumed to be entirely deposited into the asymmetric stretching mode, resulting in the following equation relating T_3 and absorbed pump energy [23]:

$$E_{abs} = \frac{N_{CO_2}(\sum_{\nu=1}^{9} E_{\nu} N_{\nu})}{\sum_{\nu=0}^{9} N_{\nu}}$$

where N_{CO_2} is the total number density of CO₂ molecules, E_v is the energy of the 00v vibrational state, and N_v is the population of this state (see Equation 2 in Chapter 2). The diamond markers in Figure 11 indicate experimentally measured T_3 values and their corresponding peak gain coefficients. It is estimated that nearly all of the pump energy is absorbed in the cell in the 200 torr case, corresponding to an absorbed energy per volume of ~60 mJ/cc. It is thus apparent that a much greater amount of pump energy is required to achieve $T_3 > 2000$ K in more than 50 torr CO₂ due to simple conservation of energy. The lack of absorbed energy in the 10 torr case is attributed to the extremely small width of the 4.3 µm CO₂ absorption spectral lines at such low pressures in comparison with the ~2 nm bandwidth of the 4.3 µm pump laser.

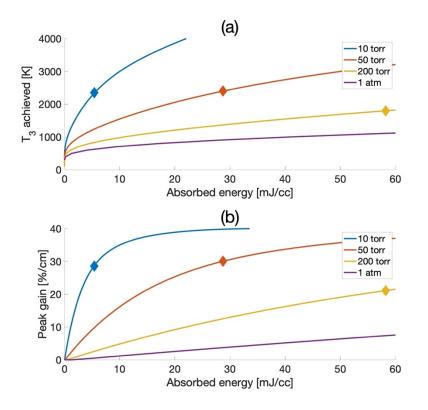


Figure 11. (a) The asymmetric stretching vibrational mode temperature T_3 and (b) the corresponding peak gain coefficient in an optically pumped CO_2 active medium vs. absorbed 4.3 µm pump energy per volume. Each color corresponds to a different amount of pure CO_2 . Experimental measurements are shown in diamonds.

To further investigate gain limitations in an optically pumped CO₂ system, measurements were performed in mixtures where the concentration of CO₂ is heavily diluted with Helium. Note that adding He serves two purposes, as it both collisionally broadens the CO₂ lines and efficiently removes heat deposited in the gas. Figure 12 shows experimental measurements of peak T_3 as a function of pressure in pure CO₂ (blue curve) and in CO₂-He mixtures containing a constant 10 torr of CO₂ and ballast He (red curve). It is clear that increasing the amount of CO₂ reduces T_3 for the reasons discussed above. It is also evident, however, that increasing the amount of Helium while maintaining constant partial pressure of CO₂ increases T_3 . This result is attributed to increased absorption of 4.3 µm pump energy due to the broadening of spectral lines via CO₂-He collisions. These results give further evidence that very high vibrational temperatures can be achieved in a CO₂ system optically pumped by Fe:ZnSe laser pulses, but a multi-atmosphere CO₂ laser requires a much more energetic pump source.

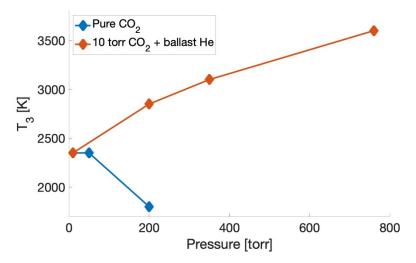


Figure 12. The peak asymmetric stretching vibrational mode temperature, T_3 , in an optically pumped CO_2 active medium as a function of total pressure for pure CO_2 (see blue curve) and for dilute CO_2 mixtures (see red curve).

4.3 High-pressure lasing

The development of the Fe:ZnSe MOPA capable of producing ≤ 60 mJ pulses allowed us to confirm whether increased pump energy could result in gain at high pressures. In addition, the tunability of the Fe:ZnSe pump laser allowed us to study the effects of varying absorption within the CO₂ active medium by changing the wavelength of the pump pulse. Again, lasing was studied using the experimental setup depicted previously in Figure 8. A 6 cm cell was filled with various mixtures of CO₂ and Helium gas and pumped at several different wavelengths with the objective of observing lasing at the highest possible total pressure.

Figure 13(a) shows the normalized absorption spectrum at ~4.3 μ m for a mixture of 0.75 atm CO₂ and 9.25 atm Helium, which was close to the optimal ratio as will be discussed below. It can be seen that varying the wavelength of the continuously tunable Fe:ZnSe pump laser along this CO₂ absorption spectrum can give access to a broad range of absorption values. The red markers in Figure 13(a) indicate the five pump wavelengths studied in this experiment, ordered from maximum (~3.5 cm⁻¹) to minimum (~0.2 cm⁻¹) absorption. Wavelengths were chosen to study a range of absorption from the peak to the wing of the distribution, and 4.256 μ m (marker 2 in Figure 13(a)) was chosen specifically because it corresponds to the fourth sub-harmonic of the wavelength of a Nd:YAG laser (1064 nm), a possible alternative pump source [30,31].

Figure 13(b) displays the maximum total pressure at which lasing was observed as a function of measured absorption coefficient for a cell filled with different partial pressures of CO_2 and ballast He. The numbers next to each marker indicate the corresponding pump wavelength as shown in Figure 13(a), and the dashed lines indicate a calculated absorption coefficient that would correspond to 50% and 99% of the pump energy being absorbed over the 6 cm cell length. Note that for each of the markers to the right of the second dashed line, the entire pump pulse energy

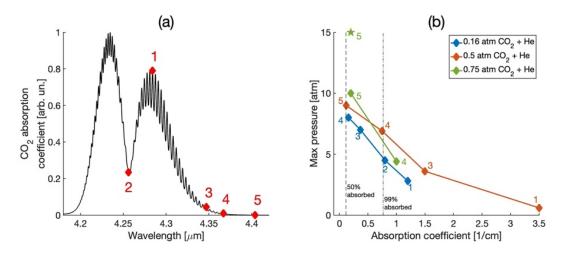


Figure 13. (a) Normalized absorption in 0.75 atm CO₂ and 9.25 atm He. Red markers indicate pump wavelengths used in experiment. (b) Maximum lasing pressure vs estimated experimental absorption coefficient. Numbers correspond to pump wavelengths indicated in (a). The green star indicates maximum lasing pressure after increasing the pump intensity by a factor of \sim 2 to \sim 5.3 MW/cm².

was absorbed. It can be seen that tuning the pump wavelength far from the peak of the absorption spectrum was necessary to achieve lasing at high pressures. This is attributed to the fact that in high absorption mixtures, nearly all of the pump energy was absorbed over a short distance, reducing the gain length of the CO_2 active medium. Increasing the partial pressure of CO_2 allowed for increased gain coefficients, but this also increased absorption and thus reduced the gain length. Note that lasing was not achieved at pressures above 5 atm when the cell was pumped at 4.256 µm. The relatively high absorption coefficient here dictates that, if using an alternate pump source limited to this wavelength, efficient short pulse amplification will require cell length optimization or utilizing a transverse pump configuration.

For high-pressure lasing in a 6 cm cell, the optimal partial pressure of CO₂ was found to be 0.75 atm and the optimal pump wavelength was found to be 4.40 μ m. Further focusing of the pump beam to increase the intensity by a factor of ~2 to ~5.3 MW/cm2 allowed for the achievement of lasing at total pressures up to 15 atm, as indicated by the green star in Figure 13(b).

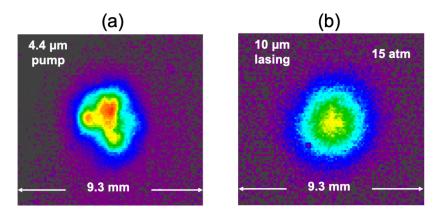


Figure 14. Beam profiles of (a) the 4.4 μ m pump laser and (b) the 10.6 μ m optically pumped CO₂ laser output at a total pressure of 15 atm.

Typical spatial beam profiles of the Fe:ZnSe pump pulse and the 10.6 μ m optically pumped CO₂ laser pulse are shown in Figure 14. Note that the maximum pressure in this experiment was ultimately limited by the gas handling system, and higher pressures can be reached with further optimization. In addition to advantageously increasing the absorption length in the cell, pumping this far from the peak of absorption eliminates the need for purging the experimental area with Ar or N₂, as CO₂ absorption in the ambient air at a wavelength of 4.40 μ m is negligible. It should also be noted that the experimentally measured absorption using high intensity, 4.40 μ m radiation was significantly higher than that theoretically predicted using HITRAN molecular constants [32]. This can be attributed to the fact that, within the cell, there may be significant absorption on the 001-002 and 002-003 absorption bands, as all vibrational levels in the asymmetric mode are populated and the absorption transitions of these sequence bands are red-shifted due to CO₂ anharmonicity. This may also indicate a non-Boltzmann distribution of population among rovibrational energy levels for highly excited CO₂ molecules and requires further study.

The optical-to-optical conversion efficiency was also measured using several different output couplers with this same optimal CO_2 pressure and pump wavelength. Figure 15(a) shows the conversion efficiency as a function of total pressure for cavities formed by a second dichroic

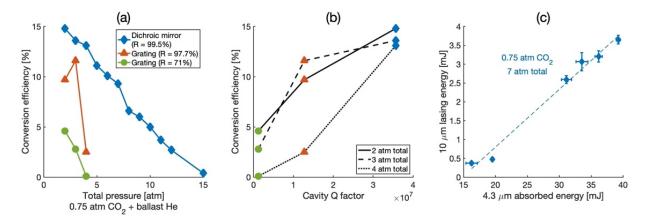


Figure 15. (a) Optical-optical conversion efficiency of the optically pumped CO₂ laser as a function of pressure using three different output couplers. (b) Conversion efficiency as a function of cavity Q factor at three different pressures.
(c) 10 μm lasing energy generated vs. 4.40 μm pump energy absorbed at 7 atm total pressure.

mirror, a 97.7% reflective diffraction grating, and a 71% reflective diffraction grating as indicated. It should be noted that flat reflective diffraction gratings were used, while the dichroic mirror had a radius of curvature of 70 cm, which likely helped reduce diffraction losses and therefore the lasing threshold. Figure 15(b) shows the same set of data, here presenting the conversion efficiency as a function of cavity Q factor. It was found that the highest reflectivity output coupler provided the best performance both in terms of maximum achievable lasing pressure and in terms of highest conversion efficiency.

Figure 15(c) shows measurements of 10.6 μ m lasing energy as a function of absorbed 4.40 μ m energy at a total pressure of 7 atm for the cavity utilizing a second dichroic mirror as an output coupler. Note that for this measurement, the range of absorbed energy was controlled not by tuning wavelength but by attenuating the pump pulse energy by a factor of 2, and small variations in the data reflect the day-to-day change of the output of the Fe:ZnSe pump laser. There is no sign of

saturation with increased pump energy, indicating the potential for scaling 10 µm output to higher energies.

4.4 Gain lifetime

Small-signal gain measurements were also performed to study the gain lifetime of an optically pumped CO₂ active medium. Relaxation rates have previously been measured in CO₂ systems pumped at ~4.3 μ m [33,34]. However, the extreme excitation of the CO₂ asymmetric stretching mode that is presented in this study may lead to different relaxation rates and has not been tested experimentally. To study this, small-signal gain and gain lifetime were measured by probing an optically pumped CO₂ system in the V-shaped double-pass scheme described previously [16].

Figure 16 shows a typical measurement of the 4.40 μ m pump pulse (in blue) and the total amplification factor of the 10.6 μ m probe pulse (in red) as a function of time for an example gas mix of 0.75 atm CO₂ and 10.25 atm He. It can be seen that, even at 11 atm, amplification reaches a maximum after the termination of the pump pulse, indicating the importance of repopulating the

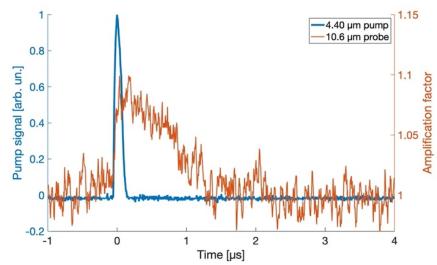


Figure 16. The 4.40 μ m pump pulse (blue) and the amplification factor of the 10.6 μ m probe pule (red) as a function of time for a gas mix of 0.75 atm CO₂ and 10.25 atm Helium.

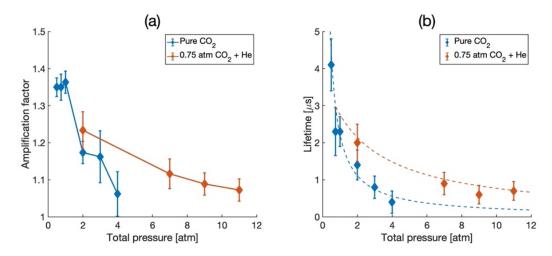


Figure 17. (a) Total amplification of the 10.6 μm probe pulse in the 6 cm optically pumped CO₂ cell and(b) gain lifetime as a function of total pressure.

upper laser level 001 by intramode collisional energy exchange to reach the maximum gain coefficient. Figure 17(a) shows the peak of this total amplification factor as a function of pressure. The blue and red curves correspond to measurements made in pure CO₂ and in 0.75 atm CO₂ mixed with ballast He, respectively. Fitting an exponential curve to the decay of these measured gain signals over time resulted in the measured gain lifetimes displayed in Figure 17(b). The dashed lines in Figure 17(b) correspond to values calculated using the experimentally determined relaxation rates for CO₂-CO₂ and CO₂-He collisions measured by Inoue et. al. and Lepoutre et. al. [33,34]. From the data in Figure 17(b), it is clear that our measured gain lifetimes agree well with these previously published values, indicating that for our pumping conditions, the physical processes occurring in an optically pumped CO₂ active medium do not differ significantly from those published in literature for a lower degree of excitation. Importantly, a long gain lifetime of ~1 µs at high pressures ≥10 atm opens the opportunity for building both regenerative and multipass amplifiers suitable for amplifying short seed pulses. Thus, in the direct optical pumping

scheme using 4.4 μ m Fe:ZnSe lasers, a high conversion efficiency of ~10% could be reached when the stored energy can be extracted in a short pulse.

4.5 Lasing in indirectly pumped CO₂

Using the ≤ 60 mJ Fe:ZnSe MOPA system, lasing was also studied in CO-CO₂-He mixtures utilizing the indirect pumping scheme, in which CO molecules are optically excited and energy is collisionally transferred to the CO₂ asymmetric stretching mode and thus the upper laser level (see Figure 2(b)). This scheme is motivated by the potential of further increasing the gain lifetime at high pressures due to the slow exchange of vibrational energy in a dual molecular system where the absorbing two-atomic molecule has a long relaxation time [12]. This scheme may also be used to avoid deleterious nonlinear refraction that can occur in CO₂ [29], which has a dipole moment three times larger than that of CO.

The same setup illustrated in Figure 8 was used for these measurements, with the Fe:ZnSe pump laser wavelength now tuned to ~4.6 μ m to coincide with absorption in the 0-1 vibrational band of CO. Figure 18(a) shows the normalized absorption spectrum of 1 atm CO, and the red markers indicate the pump wavelengths used in this experiment. Figure 18(b) shows the maximum total pressure at which 10.6 μ m lasing was analyzed for two different gas mixtures as shown. The two dashed lines again indicate the absorption coefficients at which 50% and 99% of the pump pulse energy are absorbed over a length of 6 cm, and all pump energy was absorbed for the rightmost two data points. Note that lasing at pressures above 4 atm was not achieved in either the 9:1 CO₂:CO or the 10:3:27 CO₂:CO:He mixes, and varying the absorption by tuning pump wavelength did not have an observable positive effect on the maximum lasing pressure. While we have indeed observed a gain lifetime that is several times longer than that for direct pumping, the

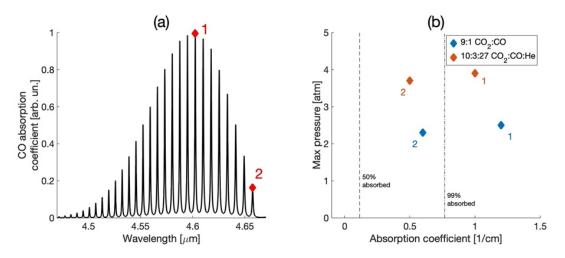


Figure 18. (a) Normalized absorption in 1 atm CO. Red markers indicate pump wavelengths used in experiment.(b) Maximum lasing pressure vs estimated experimental absorption coefficient. Numbers correspond to pump wavelengths indicated in (a).

gain values and 10 μ m output stability were clearly inferior to those measured in the case of 4.4 μ m pumping of the CO₂ medium.

For both mixtures, an audible noise corresponding to heat or pressure waves inside the cell could be heard upon the arrival of the pump pulse. The gas mixtures were flowed slowly through the cell to reduce the effects of this, but these pressure waves can play a significant role in vibrational energy exchange processes, thus hindering the performance of indirectly pumped CO_2 laser and limiting the maximum total pressure to <4 atm in this experiment. A fast flow of the gas should be able to mitigate this effect, but extra modifications were beyond the scope of our study.

4.6 Possibility of repetition rate scaling

While all the measurements above using the full Fe:ZnSe MOPA system have been conducted at a relatively low repetition rate of 3 Hz, it is important to identify the rate at which the gas molecules in the system recover from the disturbance of pressure waves caused by the absorption and

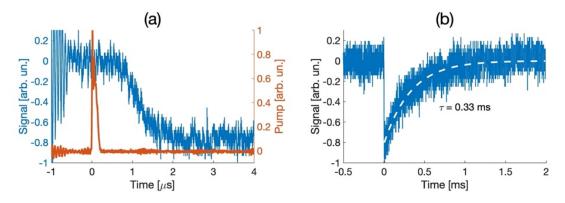


Figure 19. (a) Deflected probe signal (in blue) and Fe:ZnSe pump pulse (in red). (b) Deflected probe signal and recovery. The dashed white line indicates an exponential fit with a time constant of 0.33 ms.

dissipation of heat caused by the energetic pump pulse. To do this, we used a visible cw He-Ne probe laser to analyze how quickly the refractive index of the gas recovers after absorption of the pump pulse. Figure 19 indicates the results of these red beam deflection measurements in a gas mixture of 0.75 atm CO_2 and 9.25 atm He pumped at 4.4 μ m.

Figure 19(a) shows the probe signal in blue and the Fe:ZnSe laser pump pulse in red on a microsecond time scale. It can be seen that immediately after the ~200 ns pump pulse, the probe signal remains unaffected, but ~1 μ s after the pump pulse, the probe beam is deflected, and the signal is reduced. Figure 19(b) shows the recovery of the probe signal for a similar pump pulse energy on a longer time scale. It can be seen that after ≥ 1 ms, the gas system has returned to equilibrium and the visible probe signal returns to the initial value. Fitting an exponential curve gives a relaxation time constant of ~0.33 ms at this 10 atm total pressure. A similar result was measured in CO-CO₂ mixes, where the relaxation time constant was measured to be ~0.50 ms in a mix of 1 atm CO, 0.1 atm CO₂, and 8.9 atm He. Given that this relaxation time is inversely proportional to pressure, it can be estimated that for the high pressures (>10 atm) desirable for short-pulse amplification, the system will reset in ≤ 1 ms. This indicates that an optically pumped CO₂ laser system is capable of operating at a repetition rate of ~1 kHz.

There are currently two feasible options for the development of a 1 kHz solid-state source capable of pumping a CO₂ active medium for the amplification of 10 μ m pulses to GW level powers (with a few mJ in a few ps). A 1 kHz Fe:ZnSe laser system pumped by Er:YAG lasers providing mJ-level output has already been developed and could be scaled to provide tens of mJs utilizing commercially available 50 mJ, 1 kHz Er:YAG lasers for pumping [13]. Alternatively, the efficient frequency down-conversion of 1064 nm Nd:YAG laser pulses could be used to generate pulses at 4.256 μ m, the fourth sub-harmonic of this wavelength. Nd:YAG lasers are capable of operating at a kilohertz repetition rate, and efficient conversion of 15 ns pulses from 1064 nm to 2128 nm and from 2.1 μ m to ~3-5 μ m has already been demonstrated with optical parametric oscillators operating at or near the degeneracy point [30,31]. It should be noted that for such a CO₂ laser system producing multi-watt average powers, a certain heat exchanger similar to that used in industrial CO₂ pulsed lasers should be designed and utilized.

5 Simulation results

The demonstration of lasing and gain at high pressures up to 15 atm indicate that an optically pumped CO_2 amplifier can achieve a broad bandwidth capable of amplifying picosecond or subpicosecond pulses. Historically, however, the interest in high-pressure, optically pumped, compact CO_2 lasers was motivated by the prospect of continuous spectral tuning, and thus the experimental realization of amplifying short pulses in such a medium has never been performed or even analyzed numerically. Recently, we have shown a path towards achieving high peak powers in 10 μ m pulses by the direct amplification of a 3 ps pulse and the chirped-pulse amplification of a 1 ps pulse in a CO_2 amplifier pumped by a tunable 4.3 μ m pump source using numerical simulations [17]. These simulations were performed using the *co2amp* software program, which is based on density matrix formalism and analyzes broadband pulse amplification in a CO₂ active medium [35].

As discussed in Chapter 1, pressure broadening allows the CO₂ gain spectrum to reach a bandwidth of ~1 THz (FWHM) for the amplification of short pulses (see Figure 1). Gain narrowing limits the minimum pulse duration to ~3 ps [6], but by introducing CO₂ isotopologues with frequencies that are slightly shifted from that of the regular ${}^{12}C{}^{16}O_2$ molecule, it is possible to further increase the bandwidth of the CO₂ gain spectrum to realize the amplification of even shorter pulses. Adding the ${}^{13}C{}^{16}O_2$ isotopologue into the active medium extends the bandwidth due to the overlap between the ${}^{12}C{}^{16}O_2$ and ${}^{13}C{}^{16}O_2$ in a gas mix of 1 atm total CO₂ and 19 atm He was performed to provide an approximately constant gain over a >1 THz bandwidth. The resulting mix consisted of 63% ${}^{12}C{}^{16}O_2$ and 37% ${}^{13}C{}^{16}O_2$. Figure 20(a) shows the calculated gain profile of such a laser mix optically pumped to a vibrational temperature, *T*₃, of 3300 K. Note that

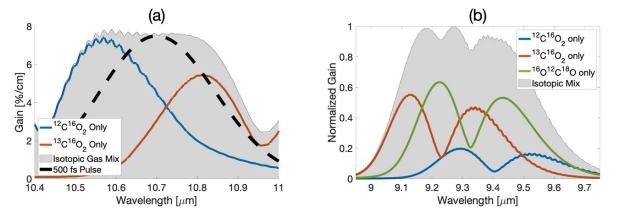


Figure 20. (a) The gain spectrum of an active medium comprised of 1 atm CO₂ (63% ¹²C¹⁶O₂ isotopologue, 37% ¹³C¹⁶O₂) and 19 atm He, pumped to a T₃ value of 3300 K. The dashed black line shows the frequency spectrum of a 500 fs seed pulse centered at 10.7 um. (b) The normalized gain spectrum of an active medium comprised of 1 atm CO₂ (21% ¹²C¹⁶O₂, 43% ¹²C¹⁸O₂, and 36% ¹²C¹⁶O¹⁸O isotopologue) and 19 atm He. A total bandwidth of ~3 THz centered around 9.3 µm wavelength is achieved.

regardless of which CO₂ isotopologue absorbs the pump radiation, near resonant interactions between the upper laser levels of the two CO₂ isotopologues quickly establish equilibrium between these levels, ensuring that a single vibrational temperature, T_3 , can be applied to both isotopologues. Later, in section 5.2, we will show the results of simulations describing the amplification of a 0.5 ps pulse in a gas mix with the gain spectrum presented in Figure 20(a).

Amplification of sub-picosecond (0.3-0.5 ps) pulses requires further broadening of the gain spectrum. The necessary bandwidth can be achieved, for example, around 9 μ m wavelength with the ¹²C¹⁸O₂ and ¹²C¹⁶O¹⁸O isotopologues. Figure 20(b) shows the normalized gain spectrum of a mix of 1 atm CO₂ isotopologues and 19 atm He. An optimized ratio of 21% ¹²C¹⁶O₂, 43% ²C¹⁶O¹⁸O, and 36% ¹²C¹⁸O₂ was used. It should be noted that while this particular gain spectrum was not used in the simulations discussed in this thesis, a similar mixture of CO₂ isotopologues at lower pressures (~10 atm) was recently used to amplify 9.2 µm, 2 ps pulses to the multi-TW level at Brookhaven National Laboratory [7].

Two simulations are presented in this chapter, both targeting the achievement of GW-level peak powers. The first considers the direct amplification of a 3 ps pulse, while the second considers the chirped-pulse amplification (CPA) of a shorter, sub-picosecond pulse. CPA reduces the intensity of the pulse via stretching before amplification, keeping the intensity of the pulse below the damage threshold of optical elements. CPA is considered only for the sub-picosecond pulse, for which a reduction of intensity becomes necessary. A length of 3 ps is chosen for the longer pulse specifically because direct amplification of 3 ps pulses in a high-pressure CO₂ amplifier is possible without measurable pulse broadening [4-6]. Only the ${}^{12}C{}^{16}O_{2}$ isotopologue is considered for this 3 ps pulse amplification, but a mixture of ${}^{12}C{}^{16}O_{2}$ and ${}^{13}C{}^{16}O_{2}$ isotopologues are considered for the case of sub-picosecond amplification (see Figure 20(a)).

Almost all other simulation parameters are kept identical for both cases. The seed pulse has an energy of 10 μ J, chosen to represent a 10 μ m pulse energy typically generated by a laser system using optical parametric amplification pumped around 1 μ m and difference frequency generation [37]. Amplification occurs in a 10 cm cell filled with 20 atm of CO₂-He (1:19) gas mix. Due to the relatively low gain and short interaction length in a CO₂ gas laser, reaching GW-level power will require the use of a regenerative amplification scheme, so the simulations consider a cell located between two mirrors with 100% reflectivity. Details on the density matrix-based model used for simulations can be found elsewhere [35].

5.1 Direct amplification of a 3 picosecond pulse

The results of a simulation of direct amplification of a 3 ps pulse centered at 10.29 μ m are shown in Figure 21. The central wavelength is chosen to correspond to the peak gain of the 10R branch of the CO₂ gain spectrum. At pressures above ~7 atm, the 10R branch is superior to the 10P branch due to the smaller frequency separation between individual rotational lines. The active medium contains 1 atm CO₂ excited such that $T_3 = 4200$ K and 19 atm He. Figure 21(a) shows the temporal profile of the pulse before and after passing through 80 cm of active medium. Over this length, a 10 μ J pulse is amplified to ~3 mJ, and the pulse length is very slightly increased to 3.1 ps. Figure 21(b) shows pulse energy as a function of amplifier length. The frequency spectrum of the final pulse is not modified, and all of the amplified energy is contained in a close to transform-limited pulse.

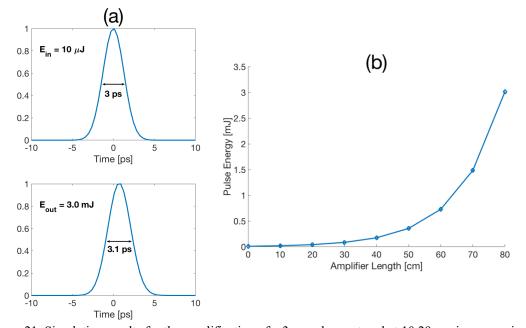


Figure 21. Simulation results for the amplification of a 3 ps pulse centered at 10.29 μ m in a gas mix of 1 atm CO₂ and 19 atm He optically excited to a T₃ value of 4200 K. (a) The temporal profile of the pulse before (top) and after (bottom) amplification. (b) The pulse energy as a function of amplifier length, discounting losses due to propagation or output coupling.

5.2 Chirped-pulse amplification of a sub-picosecond pulse

The next simulation is of the chirped-pulse amplification of a sub-picosecond pulse. As discussed earlier, amplification of a pulse of this short duration requires stretching and compressing it to avoid the damage of optical elements due to the high intensity of this pulse and to increase efficiency of interaction in a cm-scale cell. For this simulation, a 0.5 ps input pulse is considered. The pulse is stretched to 160 ps, amplified, and then compressed. A loss of 50% during compression is assumed. The amplification of sub-picosecond pulses requires gain tailoring of the active medium, which is done here with the introduction of the ${}^{13}C{}^{16}O_2$ isotopologue as described previously. For this simulation, the CO₂ medium is excited to a T_3 value of ~3300 K. The smaller

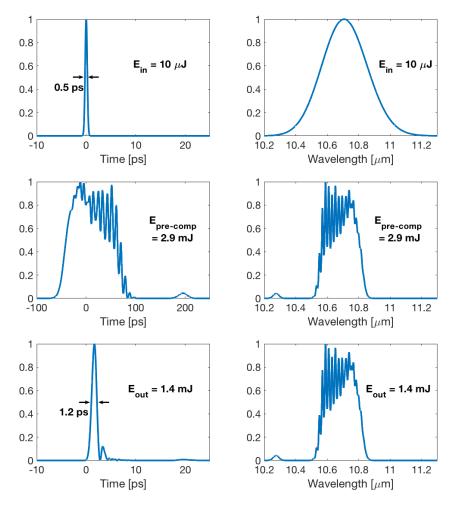


Figure 22. Simulation results for the amplification of a sub-ps pulse in an optically pumped CO₂ amplifier comprised of 1 atm CO₂ (63% ¹²CO₂, 37% ¹³CO₂) and 19 atm He. The temporal profile of the pulse before stretching (see top), after amplification (see middle), and after compression (see bottom) are shown on the left. The frequency spectrum of the pulse at each stage is shown on the right. 50% loss of energy during compression is assumed.

 T_3 value for this gas mix accounts for slightly reduced absorption of the ~4.3 µm pump pulse, a result of the smaller Einstein coefficients of the ¹³C¹⁶O₂ isotopologue [38].

Figure 22 shows the results of a simulation of CPA of a 10 μ J seed pulse centered at 10.7 μ m, where the gain is peaked. The temporal profile and frequency spectrum of the seed pulse before stretching (see top panel), after stretching and amplification (see middle panel), and after

the final compression (see bottom panel) are shown. The imperfect overlap between the frequency spectrum of the pulse and the gain spectrum of the active medium results in a longer final output pulse of 1.2 ps, but this pulse-broadening (gain-narrowing) effect is minimal and still allows for a peak power of 1 GW to be achieved, as a pulse energy of 1.4 mJ is reached after accounting for losses during compression. It is thus demonstrated that a 10 μ J, 0.5 ps pulse can be amplified to GW-level power used chirped-pulse amplification in an optically pumped CO₂ cell containing both ${}^{12}C^{16}O_2$ and ${}^{13}C^{16}O_2$ molecules.

6 Conclusion

In this thesis, lasing and gain dynamics are studied in a CO₂ active medium optically pumped at ~4.3 µm for the first time using a tunable Fe:ZnSe laser. When ~2 mJ pump pulses are used, optical-to-optical conversion efficiencies as high as 30% are demonstrated in a low-pressure (<1 atm) optically pumped CO₂ laser. The asymmetric stretching mode vibrational temperature, T_3 , and translational temperature, T, are measured as a function of time; T_3 values as high as 3600 K are measured in dilute CO₂ mixtures at a total pressure of 1 atm, and record peak gain coefficients of ~30%/cm are detected in 50 torr of pure CO2.

The use of a more energetic Fe:ZnSe MOPA system as a pump source allowed for 10 μ m lasing at total pressures up to 15 atm. It is experimentally demonstrated that tuning the pump wavelength to reduce the absorption coefficient is critical for the achievement of lasing at high pressures, and an optimum pump wavelength is found to be 4.40 μ m. A high optical-to-optical conversion efficiency of ~10% is measured at 7 atm total pressure, falling to ~5% at pressures

above 10 atm. This efficiency could be further optimized using a two-pass or transverse pumping scheme, as only \sim 70% of the total pump energy was absorbed at optimal conditions.

Gain lifetime is also measured as a function of pressure, and the $\sim 1 \ \mu s$ lifetime observed at high pressures indicate the feasibility of using optically pumped CO₂ as a multi-pass or regenerative amplifier. Simulations are performed in which a 3 ps pulse and a sub-picosecond pulse is amplified in such a regenerative amplifier to GW-level peak powers. It is shown that gain tailoring with the use of other CO₂ isotopologues can provide the broad bandwidth needed for amplifying sub-picosecond pulses. An alternative pumping scheme in which CO₂ is indirectly excited via pumping a collisional partner molecule CO is also studied, but it is determined that low gain coefficients and the generation of heat or pressure waves within the cell hinder the performance of this method. He-Ne laser probe beam deflection measurements demonstrate that a high-pressure CO₂ laser can be pumped at a maximum repetition rate of $\sim 1 \text{ kHz}$.

Our results at and above 10 atm make a compact, high-pressure optically pumped CO_2 medium a promising candidate for short 10 µm pulse amplification. With the development of even more energetic pump sources at ~4.4 µm and further work on the design optimization, such an amplification scheme may be used to progressively increase the output to TW-level power in a sub-picosecond 10 µm pulse, potentially providing a novel, compact source for high-field science and studies of atmospheric nonlinear optics in the long-wave infrared.

References

- S. Tochitsky, E. Welch, M. Polyanskiy, I. Pogorelsky, P. Panagiotopoulos, M. Kolesik, E. M. Wright, S. M. Koch, J. V. Maloney, J. Pigeon, and C. Joshi, "Megafilament in air formed by self-guided terawatt long-wavelength infrared laser," Nat. Photonics 13, 41 (2019).
- D. Haberberger, S. Tochitsky, F. Fiuza, C. Gong, R. A. Fonseca, L. O. Silva, W. B. Mori, and C. Joshi,
 "Collisionless shocks in laser-produced plasma generate monoenergetic high-energy proton beams," Nat.
 Physics 8, 95 (2012).
- 3 I. V. Pogorelsky, M. N. Polyanskiy, and W. D. Kimura, "Mid-infrared lasers for energy frontier plasma accelerators," Phys. Rev. Accel. Beams **19**, 091001 (2016).
- I. Pogorelsky, M. Babzien, I. Pavlishin, D. Stolyarov, V. Yakimenko, P. Shkolnikov, A. Pukhov, A. Zhidkov, and V. Platonenko, "Terawatt CO₂ laser: a new tool for strong-field research," Proc. SPIE 6261, 626118 (2006).
- 5 P. Corkum, "Amplification of picosecond 10 μm pulses in multiatmosphere CO₂ lasers," IEEE J. Quant. Electron. 21, 216 (1985).
- D. Haberberger, S. Tochitsky, and C. Joshi, "Fifteen terawatt picosecond CO₂ laser system," Opt. Express
 18, 17865-17875 (2010).
- 7 M. N. Polyanskiy, I. V. Pogorelsky, M. Babzien, and M. A. Palmer, "Demonstration of a 2 ps, 5 TW peak power, long-wave infrared laser based on chirped-pulse amplification with mixed-isotope CO₂ amplifiers," OSA Continuum **3**, 459 (2020).
- 8 I. Wieder, "Flame pumping and infrared maser action in CO₂," Phys. Lett. A 24, 759 (1967).
- 9 T. Y. Chang and O. R. Wood, "Optically pumped 33-atm CO₂ laser," Appl. Phys. Lett. 23, 370 (1973).
- 10 K. Stenersen and G. Wang, "Direct optical pumping of high-pressure CO₂ and N₂O lasers with a pulsed HF pump laser," IEEE J. Quant. Electron. **22**, 2236 (1986).
- B. S. Alexandrov, A. V. Aresnjev, M. A. Azarov, V. A. Drozdov, J. P. Koretsky, V. I. Mashendzhinov, V. E. Revich, and G. A. Troshchinenko, "Wide-aperture powerful high-pressure CO2 laser with optical pumping," Proc. SPIE 4644, 301 (2002).
- 12 H. Kildal and T. F. Deutsch, "Optically pumped gas lasers," in: Tunable lasers and applications, eds A. Mooradian, T. Jaeger, and P. Stokseth (Springer-Verlag, 1976), pp. 367-377.

- S. B. Mirov, I. S. Moskalev, S. Vasilyev, V. Smolski, V. V. Fedorov, D. Martyshkin, J. Peppers, M. Mirov,
 A. Dergachev, and V. Gapontsev, "Frontiers of mid-IR lasers based on transition metal doped chalcogenides," IEEE J. Sel. Top. Quant. Electron. 24, 1601829 (2018).
- 14 D. Martyshkin, K. Karki, V. Fedorov, and S. Mirov, "Room temperature, nanosecond, 60 mJ/pulse Fe:ZnSe master oscillator power amplifier system operating at 3.8-5.0 μm," Optics Express 29, 2387-2393 (2021).
- 15 D. Tovey, J. J. Pigeon, S. Ya. Tochitsky, G. Louwrens, I. Ben-Zvi, C. Joshi, D. Martyshkin, V. Fedorov, K. Karki, and S. Mirov, "Gain dynamics in a CO₂ active medium optically pumped at 4.3 μm," J. Appl. Phys. **128**, 103103 (2020).
- 16 D. Tovey, J. J. Pigeon, S. Ya. Tochitsky, G. Louwrens, I. Ben-Zvi, C. Joshi, D. Martyshkin, V. Fedorov, K. Karki, and S. Mirov, "Lasing in 15 atm CO₂ cell optically pumped by a Fe:ZnSe laser," accepted for publication in Opt. Express (2021).
- D. Tovey, S. Ya. Tochitsky, J. J. Pigeon, G. J. Louwrens, M. N. Polyanskiy, I. Ben-Zvi, and C. Joshi,
 "Multi-atmosphere picosecond CO₂ amplifier optically pumped at 4.3 μm," Appl. Optics 58(21), 5756 (2019).
- 18 W. Witteman, The CO₂ Laser (Springer-Verlag, 1987).
- 19 K. Siemsen, J. Reid, and C. Dang, "New techniques for determining vibrational temperatures, dissociation, and gain limitations in CW CO₂ lasers," IEEE J. Quant. Electron. **16**, 668 (1980).
- 20 I. M. Bertel, V. O. Petukhov, B. I. Stepanov, S. A. Trushin, and V. V. Churakov, "Investigation of the vibrational temperature kinetics in a TEA CO₂ laser," Sov. J. Quant. Electron. **12**, 1044 (1982).
- 21 B. F. Gordiets, A. I. Osipov, E. V. Stupochenko, and L. A. Shelepin, "Vibrational relaxation in gases and molecular lasers," Sov. Phys. Usp. 15, 759 (1973).
- 22 J. T. Yardley and C. B. Moore, "Intramolecular vibration-to-vibration energy transfer in carbon dioxide," J. Chem. Phys. 46, 4491 (1967).
- 23 R. C. Y. Auyeung and J. Reid, "High vibrational temperatures in optically-pumped CO2," IEEE J. Quant. Electron. 24, 573 (1988).
- 24 A. G. Maki, C. C. Chou, K. M. Evenson, L. R. Zink, and J. T. Shy, "Improved molecular constants and frequencies for the CO₂ laser from new high-J regular and hot-band frequency measurements," J. Mol. Spec. 167, 211 (1994).

- 25 R. K. Brimacombe and J. Reid, "Measurements of anomalous gain coefficients in transversely excited CO2 lasers," IEEE J. Quant Electron. 19, 1674 (1983).
- 26 C. E. Treanor, J. W. Rich, and R. G. Rehm, "Vibrational relaxation of anharmonic oscillators with exchange-dominated collisions," J. Chem. Phys. 48, 1798-1807 (1968).
- 27 V. Fedorov, D. Martyshkin, K. Karki, and S. Mirov, "Q-switched and gain-switched Fe:ZnSe lasers tunable over 3.60-5.15 μm," Opt. Express 27(10), 13934-13941 (2019).
- 28 A. S. Solodukhin, "Hot-cell-free sequence-band CO2 laser," J. Modern Opt. 34, 577 (1987).
- 29 J. J. Pigeon, D. Tovey, S. Y. Tochitsky, G. J. Louwrens, I. Ben-Zvi, D. Martyshkin, V. Fedorov, K. Karki, S. Mirov, and C. Joshi, "Resonant nonlinear refraction of 4.3-µm light in CO2 gas," Phys. Rev. A 100, 011803(R) (2019).
- 30 G. Mennerat and P. Kupecek, "High-energy narrow-linewidth tunable source in the mid infrared," in Advanced Solid State Lasers, W. Bosenberg and M. Fejer, eds. Vol 19 of OSA Trends in Optics and Photonics Series (Optical Society of America, 1998), paper FC13.
- 31 M. W. Haakestad, H. Fonnum, and E. Lippert, "Mid-infrared source with 0.2 J pulse energy based on nonlinear conversion of Q-switched pulses in ZnGeP₂," Opt. Express **22**, 8556-8564 (2014).
- 32 I. E. Gordon, L. Rothman, C. Hill, R. V. Kochanov, Y. Tan, P. F. Bernath, M. Birk, V. Boudon, A. Campargue, K. V. Chance, B. J. Drouin, J. M. Flaud, R. R. Gamache, J. T. Hodges, D. Jacquemart, V. I. Perevalov, A. Perrin, K. P. Shine, M. A. H. Smith, J. Tennyson, G. C. Toon, H. Tran, V. G. Tyuterev, A. Barbe, A. G. Császár, V. M. Devi, T. Furtenbacher, J. J. Harrison, J. M. Hartmann, A. Jolly, T. J. Johnson, T. Karman, I. Kleiner, A. A. Kyuberis, J. Loos, O. M. Lyulin, S. T. Massie, S. N. Mikhailenko, N. Moazzen-Ahmadi, H. S. P. Müller, O. V. Naumenko, A. V. Nikitin, O. L. Polyansky, M. Rey, M. Rotger, S. W. Sharpe, K. Sung, E. Starikova, S. A. Tashkun, J. V. Auwera, G. Wagner, J. Wilzewski, P. Wcisło, S. Yu, and E. J. Zak, "The HITRAN2016 molecular spectroscopic database," J. Quant. Spectrosc. Radiat. Transfer 203, 3 (2017).
- 33 G. Inoue and S. Tsuchiya, "Vibrational relaxation of CO₂(00⁰1) in CO₂, He, Ne and Ar in the temperature range of 300-140 K," J. Phys. Soc. Japan **38**(3), 870 (1975).
- 34 F. Lepoutre, G. Louis, H. Manceau, "Collisional relaxation in CO₂ between 180 and 400 K measured by the spectrophone method," Chem. Phys. Lett. **48**(3), 509 (1977).
- 35 M. Polyanskiy, "co2amp: A software program for modeling the dynamics of ultrashort pulses in optical systems with CO2 amplifiers," Appl. Optics **54**, 5136-5142 (2015).

- 36 C. Freed, L. Bradley, and R. O'Donnell, "Absolute frequencies of lasing transitions in seven CO₂ isotopic species," IEEE J. Quantum Electron. 16, 1195-1206 (1980).
- 37 D. F. Gordon, V. Hasson, H. von Bergmann, Y. Chen, A. Schmitt-Sody, and J. R. Penano, "Advanced concepts for high-power, short-pulse CO₂ laser development," Proc. SPIE **9835**, 98350Z (2016).
- 38 V. S. Starovoitov, S. A. Trushin, V. V. Churakov, and V. F. Pivovarchik, "Dipole moments of laser transitions of isotopic carbon dioxide: experiment and theory," J. Quant. Spectroscopy and Radiation Transfer 41, 153 (1989).